

Name \_\_\_\_\_

Date \_\_\_\_\_ Pd \_\_\_\_\_

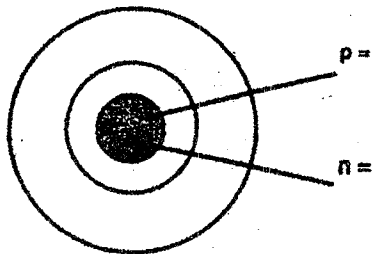
### Protons, Neutrons, Electrons

For each of the following give the Atomic Number of the element and determine the number of Protons ( $p^+$ ), Neutrons ( $n^0$ ), and Electrons ( $e^-$ ). Also determine the placement of electrons in an atom's shells.

1. Hydrogen

Atomic # \_\_\_\_\_

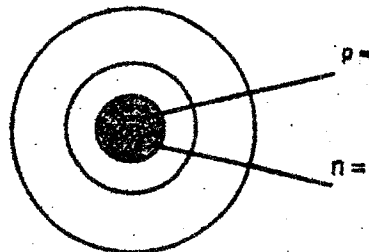
$p^+$  \_\_\_\_\_  $n^0$  \_\_\_\_\_  $e^-$  \_\_\_\_\_



2. Sodium

Atomic # \_\_\_\_\_

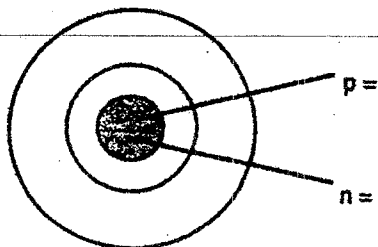
$p^+$  \_\_\_\_\_  $n^0$  \_\_\_\_\_  $e^-$  \_\_\_\_\_



3. Oxygen

Atomic # \_\_\_\_\_

$p^+$  \_\_\_\_\_  $n^0$  \_\_\_\_\_  $e^-$  \_\_\_\_\_



4. Sulfur

Atomic # \_\_\_\_\_

$p^+$  \_\_\_\_\_  $n^0$  \_\_\_\_\_  $e^-$  \_\_\_\_\_

